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Synthesis and characterization (phase, microstructure and porosity) of Li doped BCZT ceramics for Fuel Cell and clean energy technology

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ABSTRACT

In this work, we present phase and structural investigation of pure and Li (=0.08) modified perovskite structured 0.480. [Ba (Zr0.25Ti075)O3]-52(Ba0.75Ca0.25)TiO3 ceramics for versatile applications.Pure and Li modified BZT-BCT ceramic materials were synthesized by hydrothermal method. Powder X-ray diffraction studies showed single and pure rhombohedral phase formation in both pure and Li modified BZT-BCT ceramic systems. Scanning electron microscopy studies revealed that Li (=0.08) modified BZT-BCT has shown porous structure formation which is favorable microstructure features and is suitable for Fuel Cell technology and clean energy technologies as potential electrode materials.

KEYWORDS: Perovskite, Li, BZT-BCT, hydrothermal process, porous microstructure, Fuel Cell and clean energy application

1. INTRODUCTION

Ferroelectric materials with a perovskite structure have been extensively studied due to their wide applications in the fabrication of multilayer ceramic capacitors, actuators and electromechanical transducers^[1] Lead-based perovskite solid solution of Pb (Zr,Ti)O₃ (PZT) have been widely used for piezoelectric devices due to their large piezoelectricconstant^[2-4].Lead-free materials with perovskite structure such as BaTiO₃ (BT)^[5,6], Ba1-xCa_xTi1-yZryO₃ (BCZTO) ^[7-9] (Bi, Na) TiO₃ (BNT)^[10], and (Na, K) NbO₃ (NKN) ^[11]. Recently, it was reported that the addition of Li2CO₃ as a sintering additive could improve both the sinterability and the piezoelectric properties, where Li+ was doped into the lattice of BaTiO₃ during the sintering. Li₂CO₃ is a promising additive to reduce the <u>sintering</u> <u>temperature</u> for (Ba0.85Ca0.15) (Zr0.1Ti0.9)O₃ (BCZT) ceramics^[12-14]. Pure and Li modified BZT-BCT ceramic materials were synthesized by hydrothermal method. In this work, we present phase and structural investigation of pure and Li (=0.08) modified perovskite structured 0.480. [Ba(Zr0.25Ti075)O3]-52(Ba0.75Ca0.25)TiO3 ceramics for versatile applications.

2. EXPERIMENTAL

The stoichiometric pure and Li (=0.08) modified BZT-BCT ceramic powders, respectively were prepared byhydrothermalmethod using high pure reactive of Ba(OH)4, Ca(NO3)2.4H2O, ZrCl2O.8H2O, TiO2and Li2CO3. Primarily, stoichiometric ratios of reactivewere dissolved in de-ionized water under continuous magnetic stirring for 20 hours, and then, 10 M NaOH solution was mixed to the initial mixture followed by magnetic stirring the total solution for 20 hours at 25 °C. After completing the magnetic stirring, the solution was transferred to the hydrothermal reactor consisting of Teflon-lined stainless-steel autoclave, and started the hydrothermal treatment at 240° C for 24 hours, followed by cooling naturally to the room temperature. The hydrothermal reacted product was purified by washing with de-ionized water toremove sodium ions and any remanent ions, and then dried the product in an oven at 80° C to obtain pure and Li(=0.08) modified PZT-PCT ceramic powders, respectively. These powders were then mixed with 5wt% poly vinyl alcohol (PVA) binder and mixed well and compacted to circular disks (i.e., pellets) followed by sintering these green pellets at 1225°C for 3 hours and cooled to room temperature naturally, and thus obtained sintered ceramics for further phase genesis and microstructure characterizations.

The sintered ceramics were analyzed by the powder X-ray diffraction (Bruker Endeavor X-ray diffractometer model D4/MAX-Bat at a scanning rate of 0.02 °/min over a range of Bragg angles 20 between 20 and 70°) for the phase formation in the respective ceramic compositions. The apparent densities of the respective sintered samples were measured through the Archimedes method. As sintered ceramic surfaces were analyzedwith scanning electron microscopy (SEM; JEOL model JSM 840A) for the microstructure analysis.

3.RESULTS AND DISCUSSION

3.1 XRD

Figure 1 shows X-ray diffraction patters of pure and Li modified BZT-BCT ceramics. XRD results confirm the rhombohedral phase^[15]in the pure and Li modified BZT-BCT perovskite ceramics. No secondary or impurity phases were observed. All the peaks showed rhombohedral structure with homogenous phase genesis and formation. As Li was incorporated into BZT-BCT lattice, the rhombohedral phase strengthened.



Figure 1: Powder X-ray diffraction patterns of pure and Li (=0.08) modified BZT-BCNT ceramic systems

3.2 Microstructure

Figure 2 shows SEM images of pure and Li modified BZT-BCT ceramics. Microstructure studies reveal that in the pure composition grains were intact, dense and homogeneous distribution can be observed. As Lithium was introduced in the BZT-BCT system, the grains got separated forming porous structures and well-dispersed suitable for electrode materials in the Fuel Cell applications.





Figure 2: SEM micrographs of pure and Li(=0.08) modified BZT-BCT ceramic system

4 CONCLUSIONS

Pure and Li modified BZT-BCT ceramic materials were synthesized by hydrothermal method. X-ray diffraction patters of pure and Li modified BZT-BCT ceramics. XRD results confirm the rhombohedral phase in the pure and Li modified BZT-BCT perovskite ceramics. SEM images of pure and Li modified BZT-BCT ceramics. Microstructure studies reveal that in the pure composition grains were intact, dense and homogeneous distribution can be observed.

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Conflict of interest statement

Authors declare that they do not have any conflict of interest.

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